

ASYMMETRIC SYNTHESIS OF  $\alpha$ -ALKYLATED  $\alpha$ -AMINO ACIDS VIA SCHMIDT REARRANGEMENT OF  
 $\alpha, \alpha$ -BISALKYLATED  $\beta$ -KETO ESTERS

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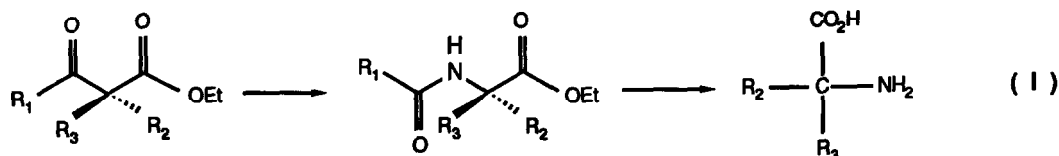
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**Abstract:**  $\alpha$ -Alkylated  $\alpha$ -amino acids are obtained in high yield and optical purity through Schmidt rearrangement of optically active  $\alpha, \alpha$ -bisalkylated  $\beta$ -keto esters.

Because of their interesting biological properties,  $\alpha$ -alkylated  $\alpha$ -amino acids have generated considerable interest both with respect to their synthesis<sup>1</sup> and their role in bioorganic chemistry.

They are known to be powerful inhibitors of those enzymes that metabolize the corresponding  $\alpha$ -unsubstituted proteinogenic amino acids.<sup>2</sup>  $\alpha$ -Alkylated  $\alpha$ -amino acids are also constituents of natural products such as the antibiotic, amicitin<sup>3</sup>, and polypeptide antibiotics<sup>4</sup> such as antiameobin I. Furthermore, this class of compounds is of importance in peptide chemistry, since introduction of an  $\alpha$ -alkylated amino acid into peptide chains restricts the available range of backbone conformations.<sup>5</sup>

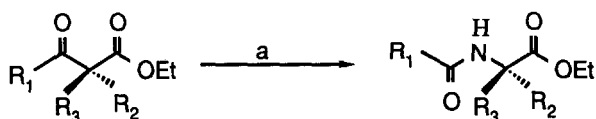
We have now been able to demonstrate<sup>6</sup> that the Schmidt rearrangement of optically active  $\alpha, \alpha$ -bisalkylated  $\beta$ -keto esters is a convenient method to generate  $\alpha$ -alkylated  $\alpha$ -amino acids in high yields and excellent optical purity (eq 1).





It is well-known that the Schmidt rearrangement of  $\beta$ -keto esters is a suitable methodology to synthesize  $\alpha$ -amino acids in high yields.<sup>7</sup> Although several studies have revealed that the Schmidt rearrangement takes place with retention of configuration,<sup>8</sup> the rearrangement of optically active  $\beta$ -keto esters has not yet been studied.

Optically active  $\beta$ -keto esters<sup>9</sup> of high enantiomeric purity have recently become available<sup>10</sup> through diastereoselective alkylation of the lithio enamines of  $\alpha$ -alkylated  $\beta$ -keto esters utilizing readily available L-valine *tert*-butyl ester as the chiral auxiliary. We synthesized several optically active  $\alpha,\alpha$ -dialkylated  $\beta$ -keto esters<sup>11</sup> and subjected them to the Schmidt rearrangement.<sup>12</sup> We obtained the corresponding *N*-acyl amino acids in excellent chemical yields (Scheme I). NMR experiments with racemic and optically active *N*-acyl  $\alpha$ -amino

Scheme I



a.  $\text{NaN}_3$ ,  $\text{CH}_3\text{SO}_3\text{H}$ ,  $\text{CHCl}_3$ , 0.5-1 h.

entry	$\text{R}_1$	$\text{R}_2$	$\text{R}_3$	yield (%)	$[\alpha]_D^{25}$	ee % **
1	$\text{CH}_3$	$\text{CH}_3$	$\text{PhCH}_2$	95	$-47.8^\circ$	>95
2	$\text{CH}_3$	$\text{CH}_3$		89	$-49.3^\circ$	>95
3		$(\text{CH}_2)_4$	$\text{PhCH}_2$	97	$+4.6^\circ$	>95
4		$(\text{CH}_2)_4$		71	$+23.9^\circ$	>95
5	$\text{CH}_3$	$\text{CH}_3$	$\text{CH}_2\text{CO}_2\text{Me}$	88	$-16.5^\circ$	70

\* The optical rotations were taken in chloroform,  $c = 1$ .

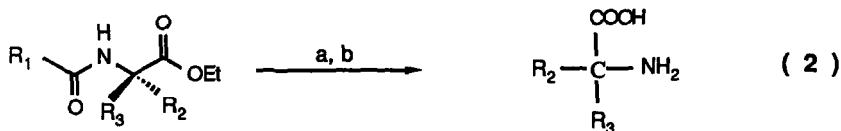
\*\* <sup>1</sup>H-NMR with chiral shift reagent (Tris[3-(heptafluoropropyl)hydroxymethylene]-d-camphorato] europium(III) derivative.

acids utilizing chiral shift reagents revealed that no detectable racemization had occurred during the Schmidt rearrangement. In the series of optically active *N*-acyl  $\alpha$ -amino acids we only observed the presence of one of the two enantiomers by NMR entry 1-4 and, therefore, assigned them as possessing an enantiomeric excess (ee) of > 95%. The  $\alpha$ -methylaspartic acid derivative (entry 5) was synthesized from the corresponding  $\alpha,\alpha$ -bisalkylated  $\beta$ -keto ester possessing an enantiomeric excess (ee) of 70%.<sup>10</sup>

With these results, we have now established for the first time that the Schmidt rearrangement of  $\alpha,\alpha$ -bisalkylated  $\beta$ -keto esters also proceeds with retention of configuration and little or no racemization.

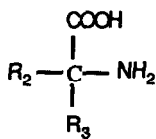
Hydrolysis of the *N*-acyl  $\alpha$ -alkylated  $\alpha$ -amino acid esters (eq 2) followed by treatment with propylene oxide in ethanol<sup>13</sup> produced the desired free D-amino acids in excellent yield and optical purity (Table 1). The ee values for D- $\alpha$ -methylphenylalanine, D- $\alpha$ -(2-naphthylmethyl)-



alanine, and D- $\alpha$ -methylaspartic acid were established by comparison with reported optical rotations.<sup>14</sup> The overall chemical yield for the three step synthesis, for example, of D- $\alpha$ -methylphenylalanine is 74%.<sup>15</sup>



a. H<sup>+</sup>. b. propylene oxide, ethanol

Table I



entry	R <sub>2</sub>	R <sub>3</sub>	yield (%)	[ $\alpha$ ] <sub>D</sub>	ee %
1	CH <sub>3</sub>	PhCH <sub>2</sub>	95	+20.5° <sup>a</sup>	97.8
2	CH <sub>3</sub>		89	+14.3° <sup>b</sup>	98.6
3	(CH <sub>2</sub> ) <sub>4</sub> CO <sub>2</sub> H	PhCH <sub>2</sub>	99 <sup>c</sup>	+3.4° <sup>b</sup>	>95
4	(CH <sub>2</sub> ) <sub>4</sub> CO <sub>2</sub> H		88 <sup>c</sup>	+8.8° <sup>d</sup>	>95
5	CH <sub>3</sub>	CH <sub>2</sub> CO <sub>2</sub> H	80	-37.7° <sup>a</sup>	69

a. The optical rotation was taken in H<sub>2</sub>O, c = 1.

b. Optical rotation in 1 N HCl.

c. Isolated as HCl salt.

d. Optical rotation in 4 N HCl.

**Acknowledgements:** Financial assistance from the NIH Biomedical Research Grant RR 5606 at the University of Kansas and from a University of Kansas New Faculty Award 3290-XX-0038 is acknowledged. Acknowledgements to Prof. D. Vander Velde for his assistance with the NMR studies and to V. Huseby for secretarial help.

## References and Notes:

- (1) For review: (a) Schoellkopf, U. In *Organic Synthesis - an interdisciplinary challenge*; Streith, J.; Prinzbach, H.; Schill, G., Eds.; Blackwell:Oxford, 1984; p. 101. (b) Seebach, D.; Imwinkelried, R.; Weber, T. In *Modern Synthetic Methods*, 1986; Scheffold, R., Ed.; Springer:Berlin, 1986; p. 125. For general reviews on the asymmetric synthesis of  $\alpha$ -amino acids: (c) Krueger, G. In *Methoden der organischen Chemie (Houben Weyl)*; Falbe, J.; Bauer, W., Eds.; Thieme:Stuttgart, 1985; Bd. E5, p. 504. (d) Barrett, G. C. In *Chemistry and Biochemistry of the Amino Acids*; Barrett, G. C., Ed.; Chapman and Hall:New York, 1985; p. 246. (e) Hoppe, D. Nachr. Chem. Techn. Lab. 1982, 30, 782. (f) Hoppe, D., Ibid. 1982, 30, 852.
- (2) For review: Jung, M. J. In *Chemistry and Biochemistry of the Amino Acids*; Barrett, G. C., Ed.; Chapman and Hall:New York, 1985; p. 227.
- (3) Hanessian, S.; Haskell, T. H. Tetrahedron Lett. 1964, 2451.
- (4) Pandey, R. C.; Meng, H.; Cook, J. C.; Rinehart, K. L. J. Am. Chem. Soc. 1977, 99, 5203.
- (5) Paul, P. K. C.; Sukumar, M.; Bardi, R.; Piazzesi, A. M.; Valle, G.; Toniolo, C.; Balaram, P. J. Am. Chem. Soc. 1986, 108, 6363.
- (6) Georg, G. I.; Guan, X.; Kant, J. Presented at the 194th National Meeting of the American Chemical Society, New Orleans, Aug. 30-Sept. 4, 1987; paper ORGN 105.
- (7) (a) Schmidt, K. F. Chem. Ber. 1924, 57, 704. (b) Schmidt, K. F. *Fortschritte in der Teerfarben und Verwandten Industrie*, 1931, 16, 2862. (c) For review: Wolff, H. *Organic Reactions*, Wiley:New York, 1946; Vol. 3, p. 307.
- (8) (a) v. Braun, J. Friehmelt, Chem. Ber. 1933, 66, 684. (b) Campbell, A.; Kenyon, J. J. Chem. Soc. 1946, 149, 25. (c) Schrecker, A. W. J. Org. Chem. 1957, 22, 33.
- (9) For review: Enders, D. In *Asymmetric Synthesis*; Morrison, J. D., Ed.; Academic Press: New York, 1984; Vol. 3, p. 317 and literature cited there.
- (10) (a) Tomioka, K.; Ando, K.; Takemasa, Y.; Koga, K. J. Am. Chem. Soc. 1984, 106, 2718. (b) Idem. Tetrahedron Lett. 1984, 25, 5677. (c) Tomioka, K.; Ando, K.; Yasuda, K.; Koga, K. Tetrahedron Lett. 1986, 27, 715.
- (11) The  $\beta$ -keto esters of entry 1 and 5 (Scheme 1) have been synthesized previously (see reference 10a). The  $\beta$ -keto esters of entry 2,3, and 4 were newly synthesized and displayed an enantiomeric excess (ee) of > 95% using NMR spectroscopy (chiral shift reagent).
- (12) A typical procedure for the Schmidt rearrangement is give below: A solution of 1.06 mmol  $\beta$ -keto ester in 5 mL of chloroform was cooled to 0°C. After the addition of 10 mmol methanesulfonic acid, sodium azide (2-4 mmol) was added in small portions. The reaction mixture was then refluxed for 0.5-1 h, quenched into ice water, neutralized with ammonium hydroxide solution and extracted with ether (3 x 5 mL). The combined organic extracts were dried over sodium sulfate, filtered, and evaporated under reduced pressure. The resulting residue was purified by column chromatography on silica gel using ethyl acetate/hexanes as eluents.
- (13) Pines, S. H.; Karady, S.; Sletzinger, M. J. Org. Chem. 1968, 33, 1758.
- (14) Schoellkopf, U.; Hartwig, W.; Groth, U.; Westphalen, K. D. Liebigs Ann. Chem. 1981, 696. Aebi, J. D.; Seebach, D. Helv. Chim. Acta 1985, 68, 1507.
- (15) All newly synthesized compounds were characterized by elemental analysis or HRMS and exhibited spectroscopic data in agreement with their structures.

(Received in USA 6 November 1987)